JAPANESE MEDICAL MATERIAL

OSVAN

(Dimethyl-lauryl-benzyl-oxyethyl-ammonium-chloride)

297200

Report No. 240

3 July 1946

MEDICAL ANALYSIS SECTION
5250th Technical Intelligence Company
APO 500

OSVAN

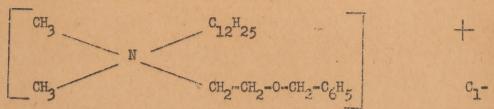
(Dimethyl-lauryl-benzyl-oxyethyl-ammonium chloride)

SOURCE: Tokyo, Japan.

IMPORTANCE: Not previously reported. An organic quaternary ammomium chloride employed as a detergent bactericide. No Identical chemical product is listed in available standard American references.

DESCRIPTION: Seven hundred and fifty bright red tablets, each containing 0.2 grams of Osvan, are contained in a green glass bottle.

SUMMARY OF GENERAL INFORMATION: Osvan is claimed to be dimethyllauryl-benzyl-oxyethyl-ammonium chloride, The following chemical constitution is recorded:



Osvan is completely water-soluble detergent bactericide. Tests record that at the end of a 5 minute period it is active in the following dilutions against the organisms listed below:

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В,	coli	1	cos	20,000
В.	typhi	1	-	40,000
В.	dysenteriae	1	tern	45,000
Sta	aphylococci	1	-	10,000

An outline of the method of manufacture has been translated from the literature furnished by the manufacturer and is part of this report.

Other information furnished by the manufacturer has also been translated and embodied in the report. This includes the results of phenol coefficient tests against various bacteria, tissue irritation tests, etc. This translation also records some information on Osvan O, a similar detergent bactericide which consists chemically of dimethyl-octyl-lauryl-ammonium chloride and is active in even greater dilutions than Osvan.

In recent years, great impetus has been given to research on antiseptics of the quaternary ammonium halide group, many of which are surface active agents. Their solutions exhibit a low surface tension, good penetrating power and fine detergent action. Evaluation of Osvan and Osvan O can only be made after a thorough literature check.

PHOTOGRAPHS:

Figure 1 - Osvan, bottle containing tablets

Figure 2 - Osvan literature

Figure 3 - Osvan literature

Figure 4 - Osvan literature

Figure 5 - Osvan literature

Figure 6 - Osvan literature

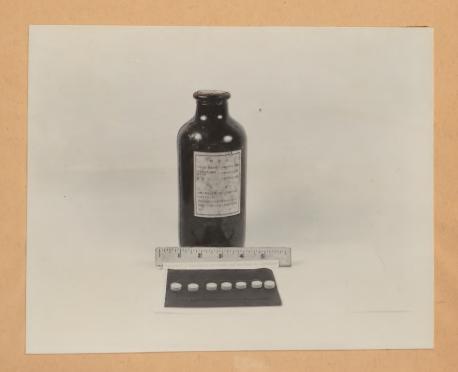
Figure 7 - Osvan literature

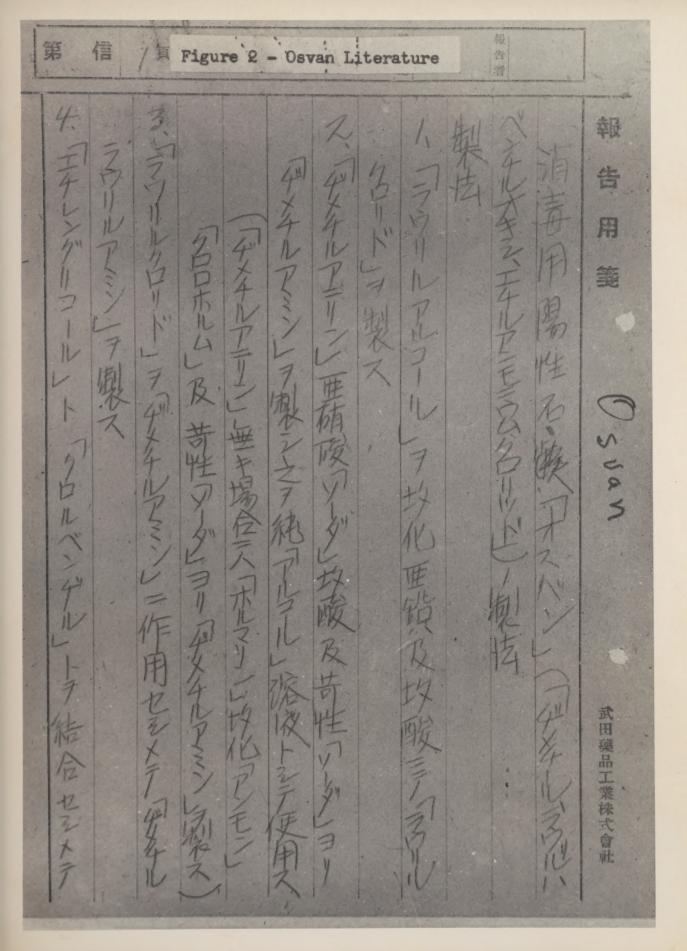
Figure 8 - Osvan literature

Figure 9 - Osvan literature

Figure 10 - Osvan literature

Figure 11 - Osvan literature





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Translation of Literature Furnished by Maruracturer

Certificate on the bacteriacidal action of inversive soap "Osvan O" - By Hideo Mori, M.D.

Preparations used in test:

"Osvan O" prepared by the Takeda Research Laboratory is dimethyl octyl lauryl ammonium chloride, clearly soluble in water and alcohol. Standard phenol and zephirol "Bayer" are used for contrasting. However, since zephirol is a liquid it is evaporated on a water bath to obtain the solid constituents and when required a fixed quantity of the dried powder is dissolved and used.

Method of Certification:

This is in accordance with the method described in the magazine of Japanese Public Health Associations.

Bacteria used in test:

Common bacilli coli

Bacilli typhoid (V Mold-Infectious Disease Research Institution)

Bacilli dysentery - general type Staphylococci, yellow (Tsuji mold)

Of the above bacteria, the staphylococci were separated from suppurations and the powerful resistant action of this mold against chemicals is shown in the following contrastive test table. Results of Experiments:

Contrastive Tests

A - Standard Phenol

		Annual registrations again to the contract of	
Type of Bacteria	Bacilli Coli	Bacilli Typhoid	Bacilli Staphylococci Dysontory
Acting Time (Min.) Strength of Preparations	2.5 5 10 15	2.5 5 10 15	2,5 5 10 15 235510 15
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B-Zebhirol

Type of Bacteria	Bacilli Coli	Bacilli Typhoid	Bacilli Dy Staphylococci Dysentery
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Type of Bacteria	Bacilli Coli	Bacilli Typhoid	Bacilli Dÿsentery	Staphylococci
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Conclusions:

From the results of the above tables the maximum dilutions of each preparation which kill the bacteria within five minutes are as follows:

Type of Bacteria Preparation	Bacilli Coli	Bacilli Typhoid	Bacilli Dysentery	Staphylococci
Phenol.	80	90	90	70
Zephirol	7500	20,000	10,000	5000
Osvan O	50,000	70,000	90,000	50,000

Comparison of each preparation against phenol whose dilution is assumed to be - 1.

Zephirol 94 222 111 71	Type of Bacteria Preparation	Bacilli Coli	Racilli Typhoid	Bacilli Dysentery	Staphylococci
0svan 0 625 778 1000 714		94 625	272	1000	77

Additional Information Composition of preparations used in test:

Three drops of each preparation of fixed aqueous dilution were dropped into the eyes of rabbits and the state of reddishness of the conjunctive and the presence of secretions were examined.

Elapsed Tin	Preparation			Zephir	ol			Osvan	0	
Tin	Dilution	1000	2500	5000	10,000	1000	2500	5000	10,000	
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Secretion	3 hours	1	+-	_	man .	+	-	Marine		

2. Mosvani, (dimethyl lauryl benzyl oxyethyl armonium chloride) is clearly soluble in water or alcohol.

The maximum dilution in which each preparation kills each type of bacteria within five minutes is shown as follows:

Type of Bacteria Preparation	Bacilli Coli	Bacilli Typhoid	Bacilli Dysentery	Staphylococci
Phenol	80	90	90 1	92
Zephirol	7500	20,000	10,000	5000
Osvan	20,000	40,000	45,000	10,000

Comparison of each preparation against a phenol dilution of I.

Type of Bacteria Preparation	Bacilli Coli	Bacilli Typhoid	Bacilli Dysentery	Staphylococci
Zephirol 0	94	222	111	71
Osvan O	625	778	1000	714
Osvan	375	444	500.	146

Translation of Literature Furnished by Manufacturer

OSVAN

Takeda Pharmaccutical Co., Ltd.
Soap for disinfection "Osvan"
Dimethyl lauryl benzyloxyethyl ammonium chloride

Method of Preparation:

1. Add zinc chloride and hydrochloric acid to lauryl alcohol to form

lauryl chloride.

2. Prepare dimethylamine from dimethylaniline, sodium nitrite, hydrochloric acid and caustic soda and dissolve it in pure alcohol. In case dimethylaniline is not available prepare dimethylamine from formalin, ammonium chloride, chloroform and caustic soda.

3. Prepare dimethyl laurylamine by the reaction of lauryl chloride and

dimethylamine.

4. Combine ethyleneglycol and benzyl chloride (use excess ethyleneglycol) to prepare benzyloxy ethyl alcohol and react this with thionyl chloride to obtain benzyloxyethylchloride.

5. Using anhydrous alcohol as a solvent and catalyzer, heat benzyloxyethylchloride and dimethyl laurylamine in the presence of urea which will result in dimethyl lauryl benzyloxy ethyl ammonium chloride. Recrystallize this with ethylacetate and purify.